

ROLE OF MULTI-SITE PHOSPHORYLATION IN REGULATION OF PIG HEART PYRUVATE DEHYDROGENASE PHOSPHATASE

Alan L. KERBEY and Philip J. RANDLE

Nuffield Department of Clinical Biochemistry, John Radcliffe Hospital, Headington, Oxford, OX3 9DU, England

Received 24 October 1979

1. Introduction

Pig Heart PDH complex is phosphorylated (with MgATP) and inactivated by PDH kinase intrinsic to the complex; and reactivated by PDHP phosphatase. Fully phosphorylated PDHP complex ($\alpha_2\text{P}_3\beta_2$) contains three phosphorylated serine residues recoverable in two tryptic phosphopeptides of known amino acid sequence [1]. Inactivation is correlated with phosphorylation of one serine residue (site 1); phosphorylation of the other two serine residues inhibits reactivation by the phosphatase [1–3]. This function of the other two sites of phosphorylation has recently been questioned [4]. We define here the optimum conditions for demonstrating different rates of reactivation of pig heart PDHP ($\alpha\text{P}.\alpha\beta_2$) and PDHP ($\alpha_2\text{P}_3\beta_2$) by ox heart or pig heart phosphatase and app. K_m values for PDHP complexes (substrate) and Mg^{2+} and Ca^{2+} (activators).

2. Experimental

2.1. Materials

Sources of biochemicals and radiochemicals are given in [1,5]. PDH complex was purified from pig

heart [6]. PDHP complexes ($\alpha\text{P}.\alpha\beta_2$ and $\alpha_2\text{P}_3\beta_2$) were prepared with $[\gamma\text{-}^{32}\text{P}]\text{ATP}$ [2]. The phosphorylated complexes (recovered by centrifugation; 90 min, $150\,000 \times g$, 4°C) were redissolved in 20 mM potassium phosphate/2 mM DTT (pH 7.0) (equiv. 30–40 units of active complex/ml) (1 unit is 1 μmol NADH formed/min at 30°C). Each preparation was dialysed (15 h, 4°C) against 2×250 ml phosphate/DTT or 50 mM potassium MOPS/2 mM DTT (pH 7.0). TCA-soluble ^{32}P was $< 0.5\%$ of total ^{32}P . Preparations of PDHP complex ($\alpha\text{P}.\alpha\beta_2$) contained $\sim 10\%$ of active complex and incorporated 0.47 ± 0.045 nmol P/unit of complex inactivated (mean \pm SEM for 4 preps.) with site occupancies corresponding to $85.3 \pm 3.28\%$ (site 1) $12.4 \pm 2.8\%$ (site 2) and $2.4 \pm 0.48\%$ (site 3) (determined by high-voltage electrophoresis of tryptic phosphopeptides [6]). Preparations of PDHP complex ($\alpha_2\text{P}_3\beta_2$) contained $\sim 3\%$ of active complex and incorporated 1.36 ± 0.115 nmol P/unit of complex inactivated (mean \pm SEM for 4 preps.) with site occupancies of $36.8 \pm 0.74\%$ (site 1), $34.7 \pm 1.03\%$ (site 2) and $28.6 \pm 1.27\%$ (site 3). Preparations of PDHP complexes were free of PDHP phosphatase. PDHP phosphatase was extracted from frozen ox or pig hearts [7] and partially purified [8]. It was freed of PDH complex by two centrifugations (60 min, $150\,000 \times g$, 4°C).

2.2. Assays and calculations

Active PDH complex was assayed spectrophotometrically by NADH formation [5]. PDHP complexes were assayed as PDH complex after complete conversion with phosphatase (in medium B, see below).

Rates of reactivation of PDHP complexes ($\alpha\text{P}.\alpha\beta_2$

Abbreviations: PDH complex, pyruvate dehydrogenase complex; PDHP, pyruvate dehydrogenase phosphate complexes; ($\alpha_2\beta_2$), tetrameric pyruvate decarboxylase (EC 1.2.4.1); PDH kinase, pyruvate dehydrogenase kinase; PDHP phosphatase, pyruvate dehydrogenase phosphate phosphatase; MOPS, 2-(*N*-morpholino) propanesulphonate; DTT, dithiothreitol; EGTA, ethanedioxybis-(ethylamine)-tetraacetate; TCA, trichloroacetic acid

and $\alpha_2\text{P}_3\beta_2$) were followed by assay of active PDH complex formed (i.e., difference from zero time). Incubations were at 30°C, (50–110 μl) and reaction initiated after 2 min preincubation by addition of substrate (or of phosphatase where K_m for substrate was determined). Incubation media were: (A) 10 mM potassium phosphate/25 mM Tris–HCl/10 mM EGTA/9.75 mM CaCl_2 /25 mM MgCl_2 ; (B) as (A) but 0.1 mM CaCl_2 and no EGTA; (C) 22.5 mM potassium MOPS/25 mM Tris–HCl/10 mM EGTA/9.75 mM CaCl_2 /25 mM MgCl_2 /1 mM potassium phosphate; (D) 47.5 mM potassium MOPS/0.1 mM CaCl_2 /25 mM MgCl_2 /1 mM potassium phosphate; (E) 44 mM potassium MOPS/0.1 mM CaCl_2 /10 mM MgCl_2 /2.5 mM potassium phosphate. All media contained 2 mM DTT and were at pH 7.0.

CaCl_2 in stock solutions and incubation media was assayed by atomic absorption spectrometry against a primary CaNO_3 standard (BDH Ltd., Poole, Dorset). Concentrations of free Ca^{2+} and Mg^{2+} were computed from dissociation constants [8].

3. Results and discussion

The results shown in fig.1,2 and table 1 were obtained with pig heart PDHP complexes and ox heart

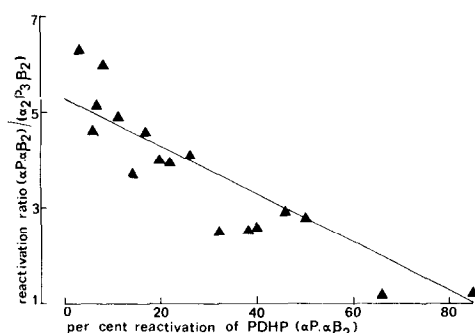


Fig.1. Incubations were made in medium A (see section 2.2); computed $[\text{Mg}^{2+}]$ 17 mM, $[\text{Ca}^{2+}]$ 13 μM , $[(\alpha\text{P}\alpha\beta_2)]$ 2.9 units/ml, $[(\alpha_2\text{P}_3\beta_2)]$ 3.0 units/ml. Samples were taken for assay of active PDH complex at times ranging from 1–20 min. There was a total of 87 obs.; values showing ($\leq 2\%$) the same percentage hydrolysis were averaged in the figure but not in the least squares linear regression analysis. Slope -0.05 ± 0.008 (mean \pm SEM for ratio/% reactivation); intercept 5.28 ± 0.293 (ratio at zero reactivation); r (correlation coefficient) -0.684 ($P < 0.001$).

phosphatase (ox heart phosphatase was more readily available and was more stable on storage than pig heart phosphatase). The results described have been duplicated with pig heart phosphatase.

Table 1
Effect of incubation medium on rates of reactivation of pig heart PDHP complex ($\alpha\text{P}\alpha\beta_2$ and $\alpha_2\text{P}_3\beta_2$) by ox heart PDHP phosphatase

Incubation medium	Ratio of rates of formation of active PDH complex; mean \pm SEM for (rate with $\alpha\text{P}\alpha\beta_2$)/(rate with $\alpha_2\text{P}_3\beta_2$) at:		
	5–20% reactivation	20–30% reactivation	30–60% reactivation
A	4.66 ± 0.25 (39)	3.56 ± 0.21 (24)	2.72 ± 0.20 (12)
B	4.77 ± 0.73 (8)	3.26 ± 0.54 (8)	2.69 ± 0.17 (11)
C	5.79 ± 1.23 (8)	3.51 ± 0.28 (5)	—
D	2.26 ± 0.19 (20) ^{a,b}	1.58 ± 0.13 (9) ^{a,b}	—
E	1.29 ± 0.20 (4) ^{a,b}	1.25 ± 0.15 (4) ^{a,b}	1.18 ± 0.03 (2)

^a $P < 0.001$ for (A–D) and (A–E)

^b $P < 0.05$ for (B–D) and (B–E)

For composition of media see section 2.2. Concentrations: (A) Mg^{2+} 16.6 mM, Ca^{2+} 13 μM , $(\alpha\text{P}\alpha\beta_2)$ 2.85 units/ml, $(\alpha_2\text{P}_3\beta_2)$ 2.90 units/ml; (B) Mg^{2+} 16.7 mM, Ca^{2+} 79 μM , $(\alpha\text{P}\alpha\beta_2)$ 3.37 units/ml, $(\alpha_2\text{P}_3\beta_2)$ 3.54 units/ml; (C) Mg^{2+} 24 mM, Ca^{2+} 15.3 μM , $(\alpha\text{P}\alpha\beta_2)$ 2.25 units/ml, $(\alpha_2\text{P}_3\beta_2)$ 2.1 units/ml; (D) Mg^{2+} 24 mM, Ca^{2+} 98 μM , $(\alpha\text{P}\alpha\beta_2)$ 2.1 units/ml, $(\alpha_2\text{P}_3\beta_2)$ 2.1 units/ml; (E) Mg^{2+} 8.2 mM, Ca^{2+} 90 μM , $(\alpha\text{P}\alpha\beta_2)$ 2.0 units/ml, $(\alpha_2\text{P}_3\beta_2)$ 2.0 units/ml. PDHP complexes added in phosphate/DTT (A,B) or MOPS/DTT (C–E). Number of observations in parentheses

3.1. Rates of reactivation of PDHP complexes ($\alpha P\alpha\beta_2$ and $\alpha_2 P_3\beta_2$); effect of extent of reactivation

As shown in fig.1, there was an inverse correlation between the ratio of reactivation ($\alpha P\alpha\beta_2$)/($\alpha_2 P_3\beta_2$) and the % reactivation of ($\alpha P\alpha\beta_2$). As analysed by least squares linear regression analysis the intercept (ratio at zero reactivation) was 5.28 ± 0.29 (mean \pm SEM). Thus the initial rate of reactivation of PDHP complex ($\alpha P\alpha\beta_2$) by phosphatase was ~ 5 -times that of fully phosphorylated PDHP complex ($\alpha_2 P_3\beta_2$). The data was generated by incubating partial and fully phosphorylated complexes for fixed times ranging from 1–20 min under identical conditions. There was a total of 87 obs. but these have been condensed by pooling in fig.1 for ease of presentation. Results with pig heart phosphatase (not shown) were very similar (intercept, 3.92 ± 0.25 ; slope, -0.07 ± 0.015 ; r , -0.985 ; 15 obs.). The results and conclusions in [2] are thus confirmed.

3.2. Effect of incubation medium on relative rates of reactivation of PDHP complexes by phosphatase

Teague et al. [4] failed to show significant differences in rates of reactivation of partial and fully phosphorylated bovine kidney PDHP complexes with bovine kidney phosphatase. MOPS buffers and 0.1 mM CaCl_2 were used in [4] whereas in [2] phosphate/Tris buffer with Ca-EGTA buffer was used. As shown in table 1 (medium E) there was no significant difference in rates of reactivation of pig heart PDHP complexes by ox heart phosphatase employing the incubation medium in [4] (i.e., the ratio of rates of reactivation was not significantly different from unity). Under all other incubation conditions investigated, the rate of reactivation of partially phosphorylated PDHP complex ($\alpha P\alpha\beta_2$) was significantly greater than that of fully phosphorylated complex ($\alpha_2 P_3\beta_2$). Increasing MgCl_2 from 10–25 mM in the incubation medium of [4] produced a modest increase in the ratio of rates of reactivation (medium D in table 1). The highest ratios were seen in phosphate/Tris buffers employing either a Ca-EGTA buffer (medium A) or 0.1 mM CaCl_2 (medium B) or in a potassium MOPS/Tris buffer with Ca-EGTA buffer (medium C in table 1). In table 1, ratios of rates of reactivation are shown for 5–20%, 20–30% and 30–60% reactivation of partially phosphorylated PDHP complex. These were generated by incubating partial and fully phos-

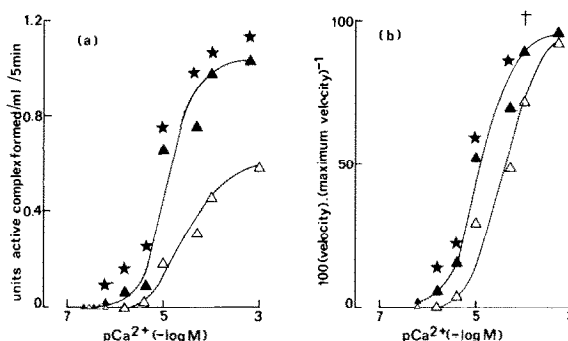


Fig.2. Incubations were made in 10 mM potassium phosphate/25 mM Tris-HCl/2 mM DTT (pH 7.0). Computed $[\text{Mg}^{2+}]$ 25 mM in all incubations; computed $[\text{Ca}^{2+}]$ as $p\text{Ca} = -\log[\text{Ca}^{2+}]$ shown on abscissa. For $p\text{Ca}^{2+}$, 6.57–4.97, CaEGTA buffers (20 mM EGTA); for $p\text{Ca}^{2+}$, 4.30–3.30, CaCl_2 . For calculation of $[\text{Mg}^{2+}]$ and $[\text{Ca}^{2+}]$ see section 2.2. Concentrations of PDHP complexes were 3.2 units/ml (\blacktriangle , $\alpha P\alpha\beta_2$) and 3.4 units/ml (\triangle , $\alpha_2 P_3\beta_2$). Samples for assay of active complex were taken after 5 min; progress curves were linear over this period (not shown). The V_{\max} values used in panel (b) were computed from the data in panel (a) and were 1.08 ± 0.71 ($\alpha P\alpha\beta_2$) and 0.63 ± 0.053 ($\alpha_2 P_3\beta_2$) (mean \pm SEM units of active complex formed/ml/5 min). Each point mean of 5 obs. * $P < 0.01$; † $P < 0.05$ for difference [$(\blacktriangle, \alpha P\alpha\beta_2) - (\triangle, \alpha_2 P_3\beta_2)$].

phorylated complexes under identical conditions for fixed times ranging from 1–20 min. The differences between the results in [4] and [2] would thus appear to be due to differences in incubation media. The negative findings with medium E appear to result from a combination of MOPS buffer, 0.1 mM CaCl_2 in place of Ca-EGTA and of 10 mM MgCl_2 in place of 25 mM MgCl_2 . Addition of KCl to 20 mM or of potassium phosphate to 9 mM in medium E did not increase the ratio (not shown). Addition of MOPS to 50 mM in medium A did not change the ratio (not shown).

3.3. Effect of concentrations of PDHP complexes, Mg^{2+} and Ca^{2+} on rates of reactivation

In these experiments active PDH complex was assayed after 5 min incubation (control experiments showed linear progress curves for 5 min at highest rates of reactivation). The app. K_m for both PDHP complexes was 22.2 units/ml (as equivalent PDH complex); V_{\max} values were 3.03 ($\alpha P\alpha\beta_2$) and 0.81 ($\alpha_2 P_3\beta_2$) (in units PDH complex formed/ml incubate/

5 min). The ratio of V_{\max} values was thus 3.7 (medium A; 4 concentrations of PDHP complexes, range 2.3–25.2 units/ml; 5 obs. each concentration). The app. K_m for Mg^{2+} was 0.59 mM (both PDHP complexes) and V_{\max} values were 1.56 ($\alpha P.\alpha\beta_2$) and 0.84 ($\alpha_2 P_3\beta_2$). The ratio of V_{\max} values was 1.86 [medium B; four Mg^{2+} values over 0.64–10 mM; PDHP ($\alpha P.\alpha\beta_2$) 2.76 units/ml, PDHP ($\alpha_2 P_3\beta_2$) 3.56 units/ml; 10 obs. at each Mg^{2+} concentration].

The effect of Ca^{2+} concentration on rates of reactivation is shown in fig.2(a). Neither preparation of PDHP complex showed significant reactivation in the absence of added Ca^{2+} . The concentrations of Ca^{2+} required for significant reactivation were 0.64 μM ($\alpha P.\alpha\beta_2$) and 3.8 μM ($\alpha_2 P_3\beta_2$). The V_{\max} values (computed by the method in [9]) were 1.08 ± 0.071 ($\alpha P.\alpha\beta_2$) and 0.63 ± 0.053 ($\alpha_2 P_3\beta_2$) (mean \pm SEM for units PDH complex formed/ml/5 min; P for difference < 0.001). The ratio of V_{\max} values was 1.71 ± 0.14 ($P < 0.001$ for difference from unity). Computed app. K_m values were $15.6 \pm 3.45 \mu M$ ($\alpha P.\alpha\beta_2$) and 43.8 ± 11.42 ($\alpha_2 P_3\beta_2$). The difference is not statistically different. As shown in fig.2(b) the observed values of (rate of reactivation) as a fraction of V_{\max} at each Ca^{2+} concentration below 500 μM were significantly greater for ($\alpha P.\alpha\beta_2$) than for ($\alpha_2 P_3\beta_2$). This would appear to show that the app. K_m for Ca^{2+} for reactivation of ($\alpha P.\alpha\beta_2$) by ox heart phosphatase is lower than for reactivation of ($\alpha_2 P_3\beta_2$). These experiments have been repeated with pig heart phosphatase with closely similar results (V_{\max} values 0.42 ± 0.059 and 0.24 ± 0.05 ; app. K_m values $26.6 \pm 12.6 \mu M$ and $116.1 \pm 64.3 \mu M$ Ca^{2+}).

The app. K_m values for Ca^{2+} in the present study are much higher than the value of $\sim 0.7 \mu M$ Ca^{2+} obtained in [10] but the app. K_m for Mg^{2+} was similar. This increase in K_m for Ca^{2+} is borne out by the higher app. K_m for PDHP complex in the present study (22 units/ml at 13 μM Ca^{2+}) (2.5 units/ml at 10 μM Ca^{2+} in [10]). Ca^{2+} lowers the app. K_m for the PDHP complex [10,11]; the rise in app. K_m for Ca^{2+} should therefore increase the app. K_m for PDHP complex at the Ca^{2+} concentration employed. The higher app. K_m for Ca^{2+} in the present study is attributed to the new method of PDH complex purification [6] used. Depletion of PDHP complexes of Ca^{2+} increases the app. K_m for Ca^{2+} [10] and the new method of preparation of PDH complex (and hence

of PDHP complex) [6] should lead to greater depletion of Ca^{2+} than the method used in [5,10]. The studies [2] which first showed that phosphorylation of sites 2 and 3 in the pig heart PDHP complex inhibits reactivation by phosphatase used PDH complex prepared by the old method [5] and showed the expected higher sensitivity to Ca^{2+} in the phosphatase reaction.

This study has shown that the inhibitory effect of site 2 and site 3 phosphorylations on reactivation of PDHP complex by phosphatase is demonstrable at V_{\max} for PDHP complexes (substrates) and for Mg^{2+} or Ca^{2+} (activators). It has been shown that phosphorylation of sites 2 and 3 specifically increases app. K_m for Ca^{2+} thus emphasising the potential significance of mitochondrial Ca^{2+} concentration as a regulator of PDH complex activity in vivo. It would seem important to establish the mitochondrial concentration of Ca^{2+} and the K_m for Ca^{2+} of the phosphatase reaction in mitochondria.

Acknowledgements

We thank Mrs P. M. Radcliffe for skilled technical assistance, Mr D. Quantrell for calcium estimations and the Medical Research Council and British Diabetic Association for their support.

References

- [1] Sugden, P. H., Kerbey, A. L., Randle, P. J., Waller, C. A. and Reed, K. B. M. (1979) *Biochem. J.* 181, 419–426.
- [2] Sugden, P. H., Hutson, N. J., Kerbey, A. L. and Randle, P. J. (1978) *Biochem. J.* 169, 433–435.
- [3] Sugden, P. H. and Randle, P. J. (1978) *Biochem. J.* 173, 659–668.
- [4] Teague, W. M., Pettit, F. H., Yeaman, S. J. and Reed, L. J. (1979) *Biochem. Biophys. Res. Commun.* 87, 244–252.
- [5] Cooper, R. H., Randle, P. J. and Denton, R. M. (1974) *Biochem. J.* 143, 625–641.
- [6] Kerbey, A. L., Radcliffe, P. M., Randle, P. J. and Sugden, P. H. (1979) *Biochem. J.* 181, 427–433.
- [7] Siess, E. A. and Wieland, O. H. (1972) *Eur. J. Biochem.* 26, 96–105.
- [8] Severson, D. L., Denton, R. M., Pask, H. T. and Randle, P. J. (1974) *Biochem. J.* 140, 225–237.
- [9] Jones, A. (1970) *Comput. J.* 13, 301–308.
- [10] Randle, P. J., Denton, R. M., Pask, H. T. and Severson, D. L. (1974) *Biochem. Soc. Symp.* 39, 75–87.
- [11] Pettit, F. H., Roche, T. E. and Reed, L. J. (1972) *Biochem. Biophys. Res. Commun.* 49, 563–571.